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- Laboratory and safety

The Matter and Energy section covers:


- Organization of matter
- Particulate structure of matter
- Chemical/physical properties/changes
- Conservation of energy and matter
- Forms of energy
- Temperature and thermal energy
- Phase transitions
- Kinetic molecular theory and ideal gas laws
- Energetics of chemical reactions / laws of thermodynamics
- Current model of atomic structure / electron configuration
- Electronic absorption and emission spectra
- Naming simple inorganic compounds
- Naming common organic compounds
- Mole Concept
- Bonding and structure
- Periodic table and trends

The Chemical Reactions section covers:

- Radioactivity
- Balancing chemical equations and stoichiometry
- Chemical kinetics
- Chemical reaction equilibrium

- Oxidation-reduction reactions
- Important biochemical compounds
- Common organic compounds
- Solutions and solubility
- Equilibrium in ionic solutions
- Properties of acids and bases
- pH scale
- Acid-base titrations
- Equilibrium relationships in acid-base chemistry

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